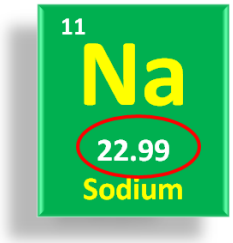
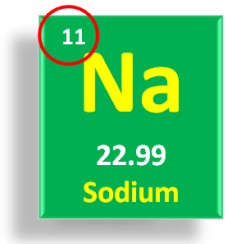
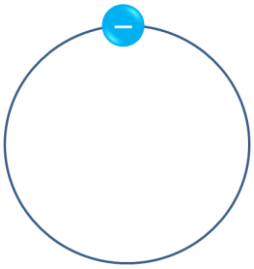
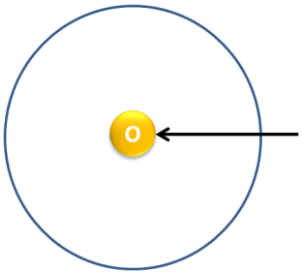
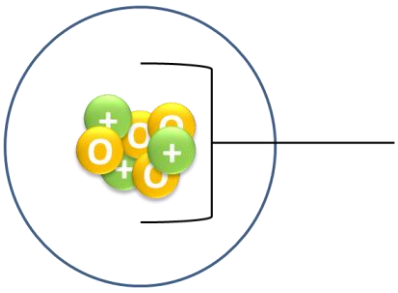
Name: _____ Period:_____ Date:_____

Atomic Structure Vocabulary – Teacher Edition

Hint	Key Term	Definition or Example
	Atomic mass	
	Atomic number	
	Electron	
	Neutron	
	Nucleus	

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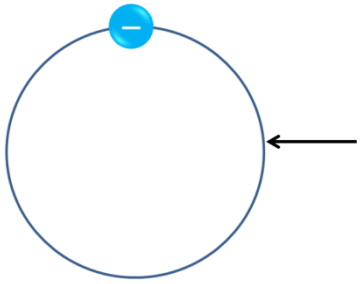
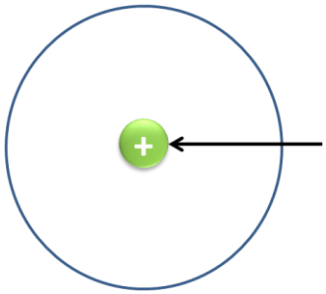
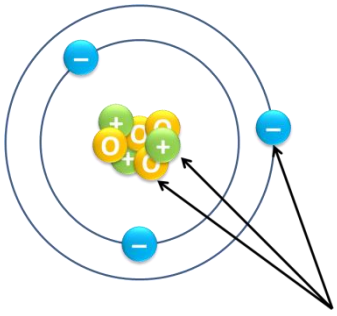
	**Orbital**	
	**Proton**	
	**Subatomic particles**	

Name: _____ Period:_____ Date:_____

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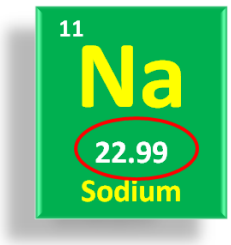
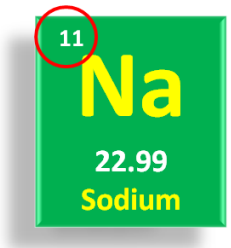
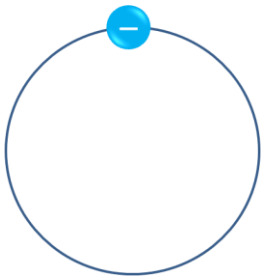
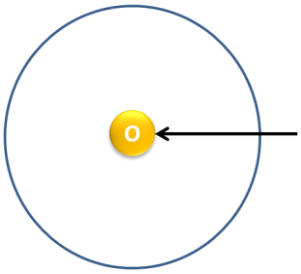
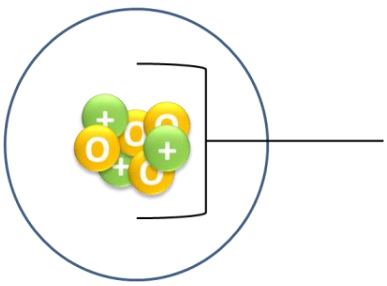
Hint	Key Term	Definition or Example
	**Atomic mass**	The number of protons contained in an atom
	**Atomic number**	The number of protons and neutrons contained in the nucleus of an atom.
	**Electron**	A small, negatively charged particle which orbits the nucleus of an atom
	**Neutron**	Heavy subatomic particle found in the nucleus of the atom which has no charge.
	**Nucleus**	The central area within an atom which contains the neutrons and protons.

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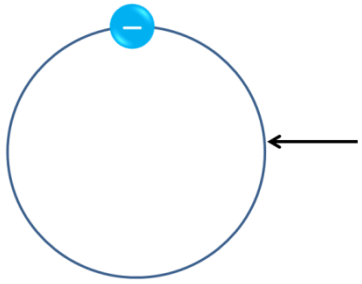
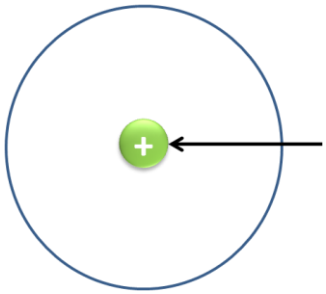
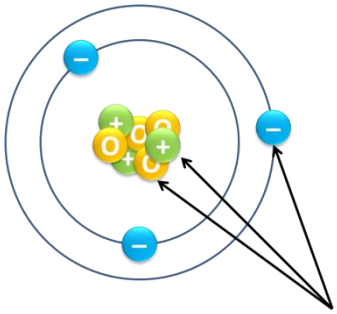
	**Orbital**	Energy level surrounding the nucleus which is occupied by electrons.
	**Proton**	Heavy subatomic particle found in the atom's nucleus which has a positive charge.
	**Subatomic particles**	Particles found within or in orbit around an atom which determine its chemical properties.